

Date: 2023 Feb 4

①

①  $PV = kT$

Critical temperature

$$\left(P + \frac{a}{V^2}\right)(V-b) = kT$$

$$a = kT_c b$$

②  $P = \frac{kT}{V-b} - \frac{a}{V^2}$

$$= \frac{kT}{V-b} - \frac{kT_c}{b} \left(\frac{b}{V}\right)^2$$

$$\frac{P}{\left(\frac{kT_c}{b}\right)} = \frac{kT}{kT_c} \frac{1}{\left(\frac{V}{b} - 1\right)} - \left(\frac{b}{V}\right)^2$$

③  $\tilde{P} = \frac{P}{kT_c/b} \quad \tilde{T} = \frac{kT}{kT_c} \quad \tilde{V} = \frac{V}{b}$

$$\tilde{P} = \frac{\tilde{T}}{\tilde{V}-1} - \frac{1}{\tilde{V}^2}$$

$V \rightarrow \infty \quad \tilde{P} = \frac{\tilde{T}}{\tilde{V}}$   
ideal gas

$V \rightarrow 1+ \quad \tilde{P} = \frac{\tilde{T}}{(\tilde{V}-1)}$

④  $\frac{\partial \tilde{P}}{\partial \tilde{V}} = -\frac{\tilde{T}}{(\tilde{V}-1)^2} + \frac{2}{\tilde{V}^3}$

$$\frac{\partial \tilde{P}}{\partial \tilde{V}^2} = 0 \quad -\tilde{T} \tilde{V}^3 + 2(\tilde{V}-1)^2 = 0$$

$$\tilde{T} \tilde{V}^3 - 2\tilde{V}^2 + 4\tilde{V} - 2 = 0$$

5) Cubic equation

$$ax^3 + bx^2 + cx + d = 0$$

$$\Delta = 18abcd - 4b^3d + b^2c^2 - 4ac^3 - 27a^2d^2$$

$\Delta > 0 \Rightarrow$  three real roots

$\Delta < 0 \Rightarrow$  one real root

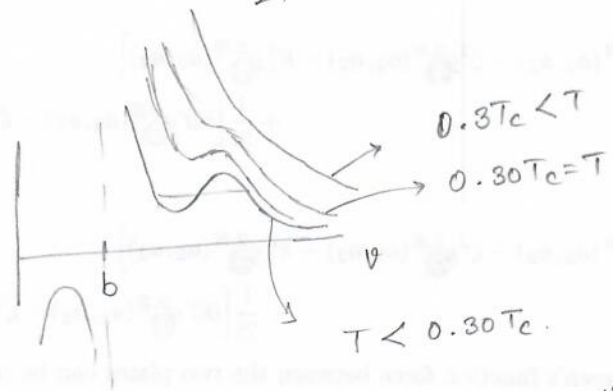
6)  $a = \tilde{t} \quad b = -2 \quad c = +4 \quad d = -2$

$$\begin{aligned} \Delta &= 18 \tilde{t} (-2)(4)(-2) - 4(-2)^3(-2) + (-2)^2(4)^2 - 4\tilde{t}(4)^3 - 27\tilde{t}^2(-2)^2 \\ &= 288\tilde{t} - 64 + 64 - 256\tilde{t} - 108\tilde{t}^2 \\ &= 32\tilde{t} - 108\tilde{t}^2 \\ &= \tilde{t}(32 - 108\tilde{t}) = 4\tilde{t}(8 - 27\tilde{t}) \end{aligned}$$

$\Delta < 0 \Rightarrow \frac{8}{27} < \tilde{t}$   
 $\frac{8}{27} kT_c < kT$

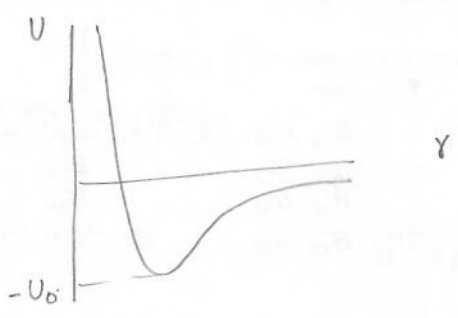
$$\frac{8}{27} = 0.30$$

7)



The third root is for  $v < b!$

8



9

$$Z(P, V, T) = \frac{1}{\Lambda^{3N}} \frac{1}{N!} \int d^3r_1 \dots \int d^3r_N e^{-\frac{1}{kT} \sum_{i < j} U(|\vec{r}_i - \vec{r}_j|)}$$

↙ length

Probability that the system is at position  $\vec{r}_1, \vec{r}_2, \dots, \vec{r}_N$

$$= \frac{\frac{1}{\Lambda^{3N}} \frac{1}{N!} e^{-\frac{1}{kT} \sum_{i < j} U(|\vec{r}_i - \vec{r}_j|)}}{Z}$$

10

$$Z = \frac{V^N}{N!} e^{+N \frac{k b_0}{kT}} = \left(\frac{V}{N}\right)^N e^{N \frac{U_0}{kT}}$$

$$F = -kT \ln Z = -kT \left[ N \ln V + N \frac{U_0}{kT} \right]$$

$$= -NkT \ln V - N U_0$$

$$S = -\frac{\partial F}{\partial T} = Nk \ln V$$

$$P = -\frac{\partial F}{\partial V} = NkT \frac{1}{V} \frac{1}{N} = \frac{kT}{V}$$

⑪ van der Waals

$$Z = (v-b)^N e^{N \frac{U_0}{kT} \frac{kT_c}{U_0} \frac{b}{v}}$$
$$= (v-b)^N e^{N \frac{U_0}{kT} \frac{a}{v}}$$

$$\frac{U_0}{kT} \rightarrow \frac{U_0}{kT} \frac{U_0}{U_0} \frac{1}{v}$$

$$U_0 v_0 = a$$
$$= kT_c b$$

$$F = -kT \ln Z$$
$$= -kT \left[ N \ln(v-b) + N \frac{U_0}{kT} \frac{a/v_0}{} \right]$$
$$= -NkT \left[ \ln(v-b) + \frac{kT_c}{kT} \frac{b}{v} \right]$$

$$S = -\frac{\partial F}{\partial T} = Nk \ln(v-b)$$

$$F = -NkT \ln(v-b) - NkT_c \frac{b}{v}$$

$$P = -\frac{\partial F}{\partial v} = \frac{NkT}{(v-b)} \frac{1}{N} + NkT_c \frac{b}{v^2} \frac{(-1)}{N}$$
$$= \frac{kT}{v-b} - kT_c \frac{b}{v^2}$$

⑫ van der Waals

①  $v \rightarrow v-b$  (excluded volume)

②  $\frac{U_0}{kT} \rightarrow \frac{U_0}{kT} \frac{U_0}{U_0} \frac{1}{v} = \frac{kT_c}{kT} \frac{b}{v}$